

Why the Lewis number seems to have an unexpected importance even in highly turbulent flames !

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with support from

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Premixed Flame Workshop - Montreal - 09. August 2008

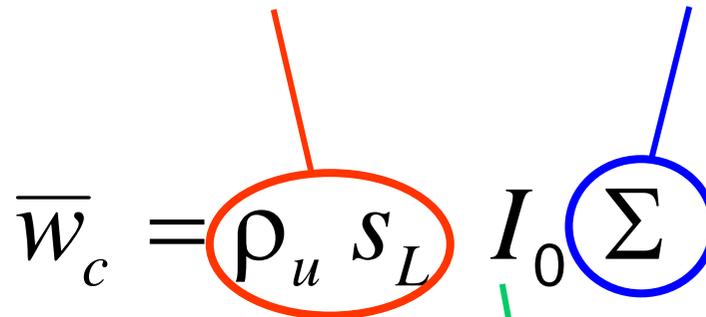
Flamelet-Ansatz für Vormischflammen

Flamelet-Idee:

- Gefaltete laminar-ähnliche Flamme
- Flamme beschrieben durch mittleren Reaktionsfortschritt \bar{c}
 \bar{c} = Wahrscheinlichkeit, verbranntes Gas zu finden
- 1 Transportgleichung für mittleren Reaktionsfortschritt (statt N Speziesgleich.)

$$\frac{\partial}{\partial t} (\bar{\rho} \bar{c}) + \frac{\partial}{\partial x_i} (\bar{\rho} \bar{u}_i \bar{c}) = \frac{\partial}{\partial x_i} \left(\bar{\rho} \frac{v_t}{Sc} \frac{\partial \bar{c}}{\partial x_i} \right) + \bar{w}_c$$

- Reaktionsrate wie im **laminaren Fall** x **Faltungsfaktor Σ**

$$\bar{w}_c = \rho_u s_L I_0 \Sigma$$


Korrektur
(Flammenkrümmung)

Aber Modell für Σ
nötig ...

Ansatz: (Zimont, Lipatnikov, 1995)

$$\overline{w}_c = \rho_u s_L I_0 \Sigma$$

$$\hat{s}_T \cdot |\nabla \overline{c}|$$

Turbulente
Brenngeschwindigkeit

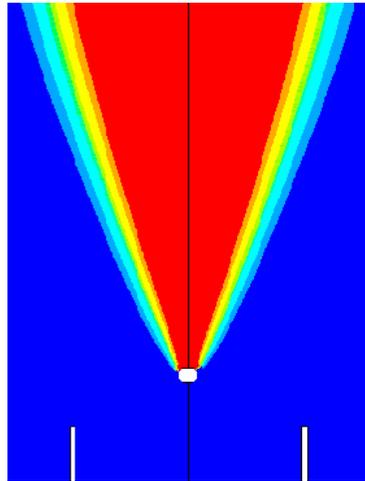
$$\hat{s}_T = f(\text{Re}_t, u', s_L, \dots)$$

Ort der Reaktionszone
(statistisch gemittelt)

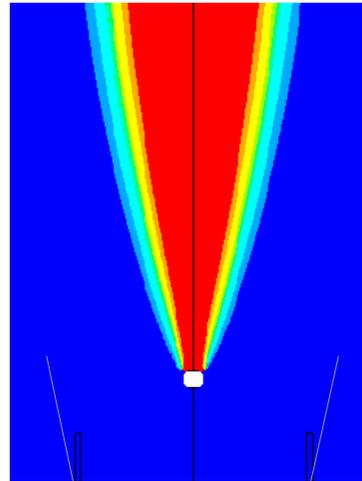
Turbulenzeinfluss modelliert
mit algebraischer Beziehung

Lösung als Subroutine mit Standard-CFD-Programm (Fluent)

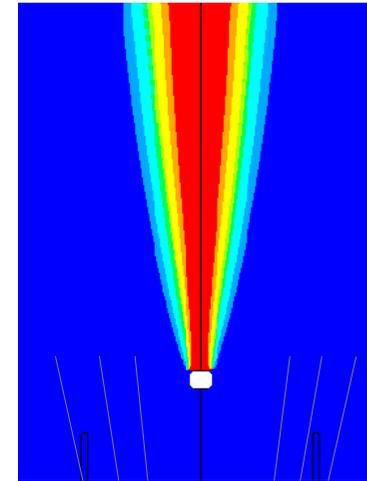
Simulation
(S. Hölzler)



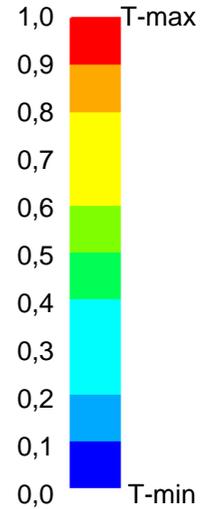
$\lambda = 1.39$



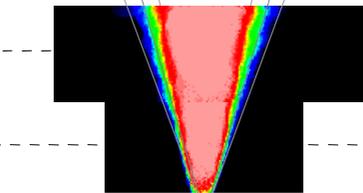
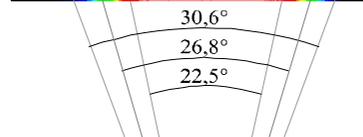
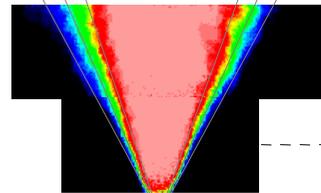
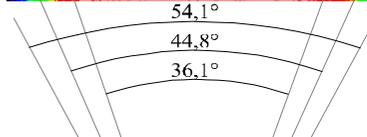
$\lambda = 1.68$



$\lambda = 1,95$



Experiment
2-dim. Rayleigh-
Thermometry
(A. Soika)



$\dot{m} = 18 \text{ kg/h}$

Bunsen Flame, Premixed

$p = 1 - 10$ (30) bar

$D = 20$ mm

$U = 2 \dots 4.6$ m/s

$Re = 2\,500 \dots 50\,000$

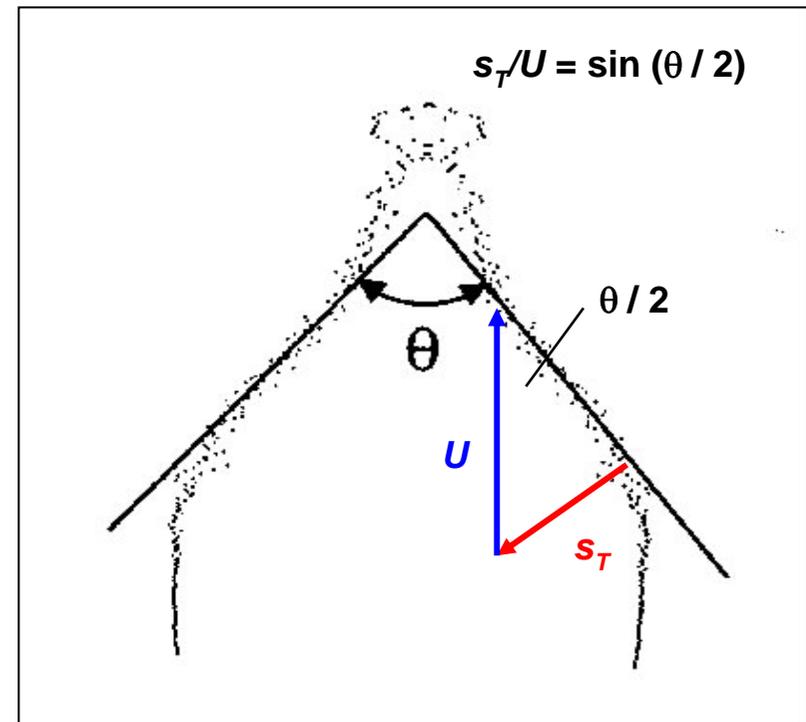
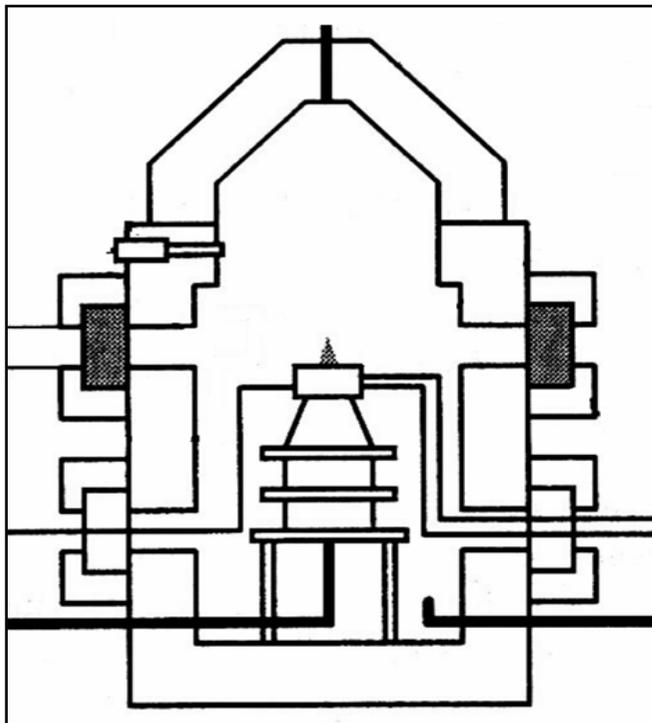
u', l_x measured (centre of inlet)

Methane/Air $\Phi = 0.9$

Ethylene/Air $\Phi = 0.5, 0.7, 0.9$

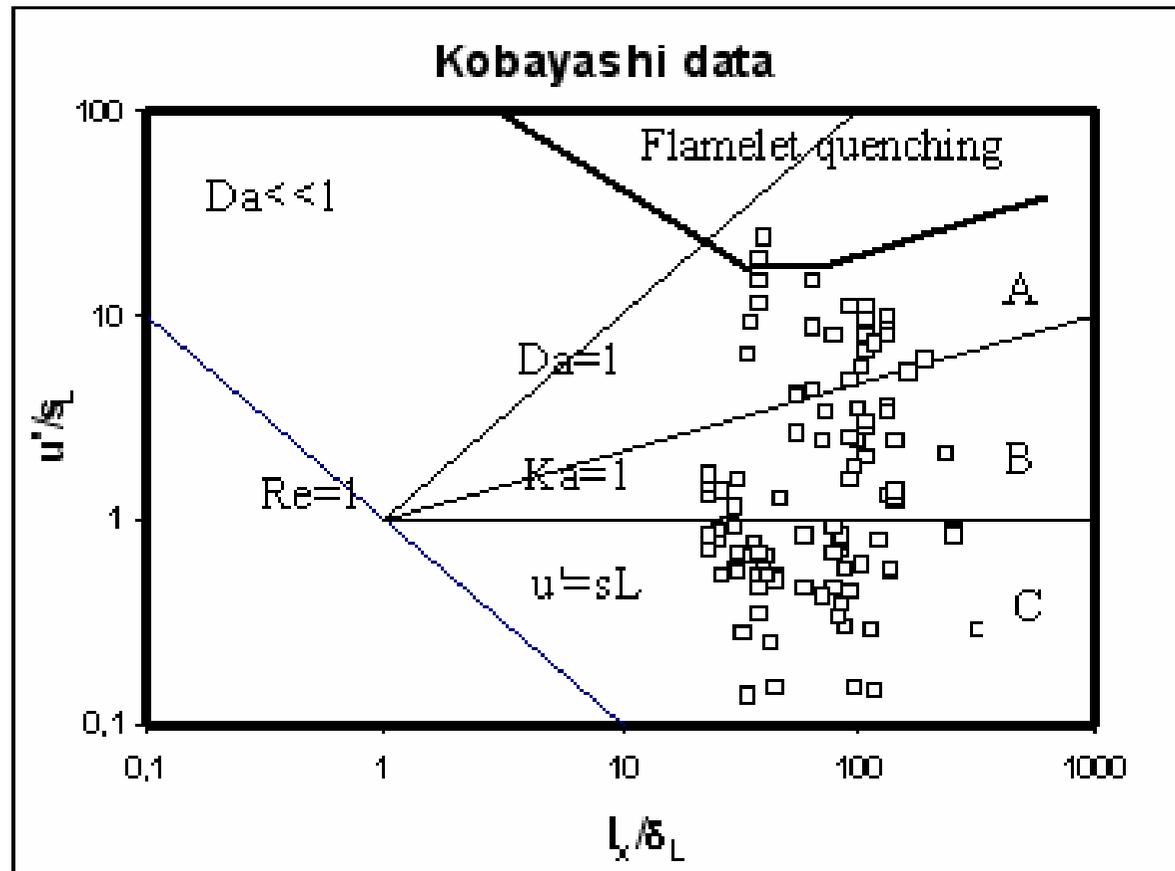
Propane/Air $\Phi = 0.9$

Data of 101 different flames



Test Data Series (Kobayashi)

fills broad range in Borghi-Peters Regime diagram



Data of 101 different flames

$\rho = 1 - 10$ bar

Methane/Air

Φ

= 0.9

Ethylene/Air

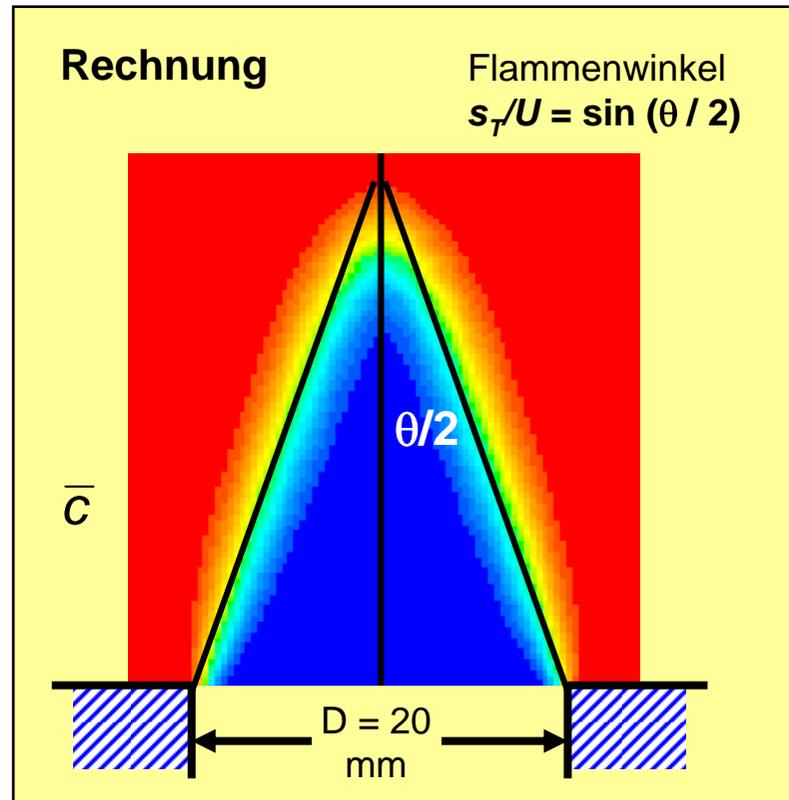
Φ

= 0.5, 0.7, 0.9

Propane/Air

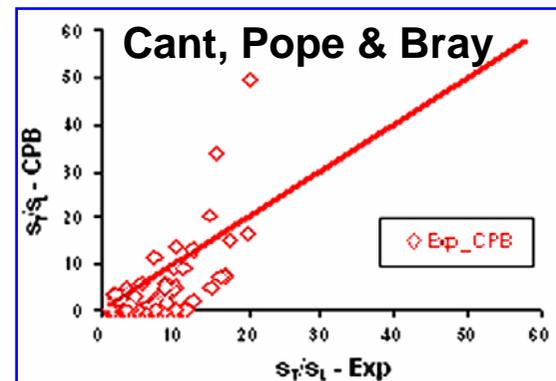
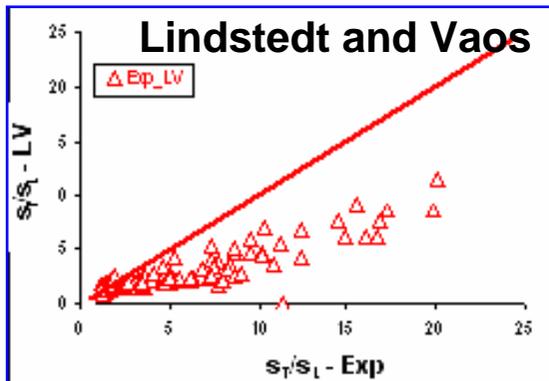
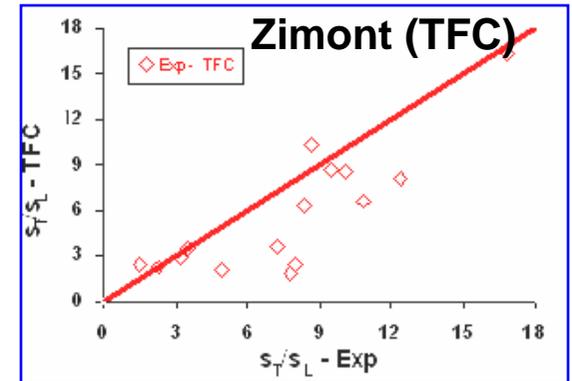
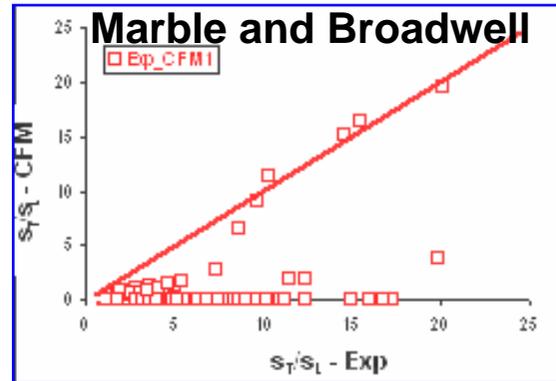
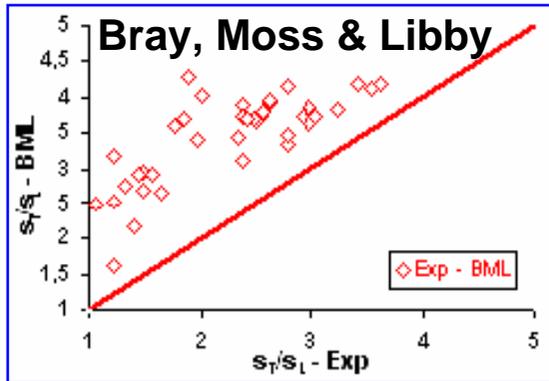
Φ

= 0.9



Calculation with different models

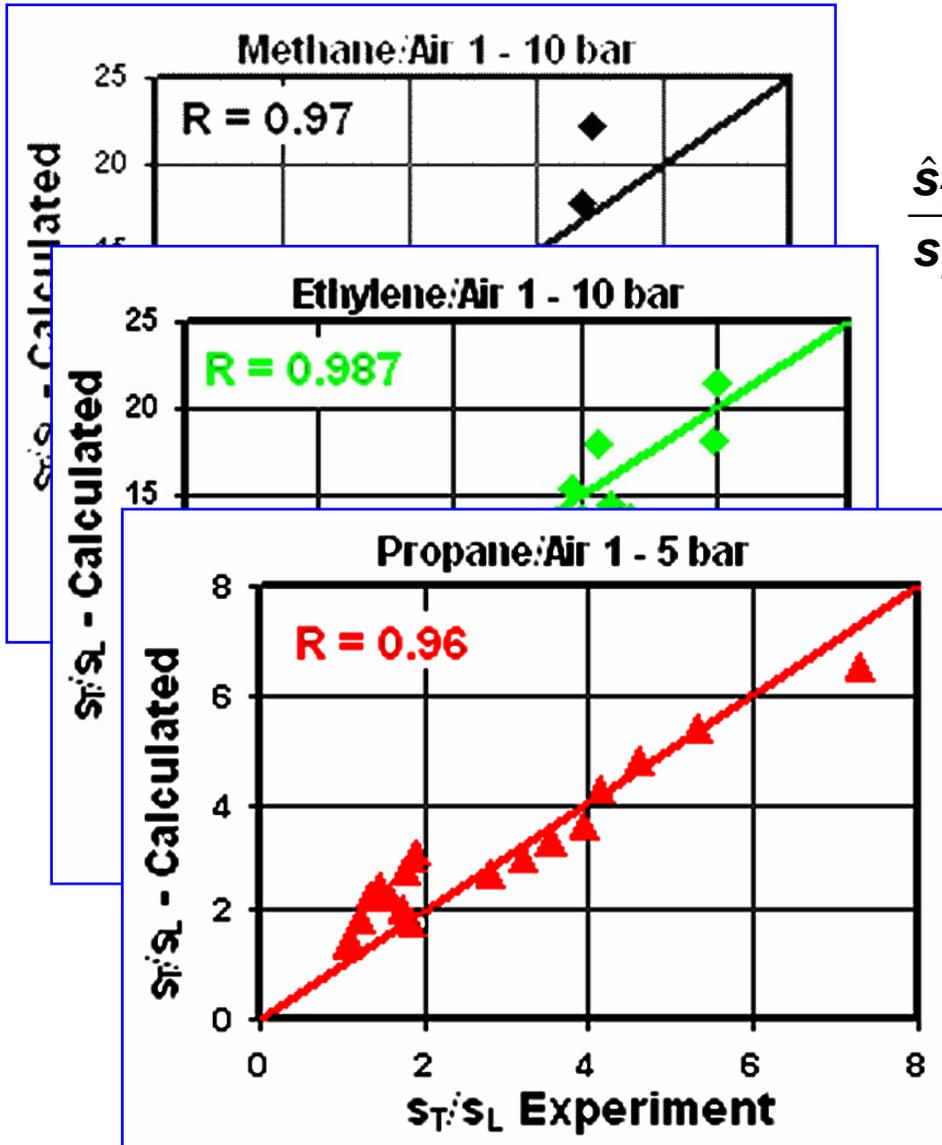
Calculated Flame Angle vs. measured Flame Angle
(each dimensionless)



Algebraic BML-like models

Flame surface density models

Algebraic gradient approach model



Methan-Luft, 1 - 10 bar, 17 Flammen

$$\frac{\hat{s}_T}{s_L} = \frac{A_T}{A_L} = 1 + 0.46 \cdot \text{Re}_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

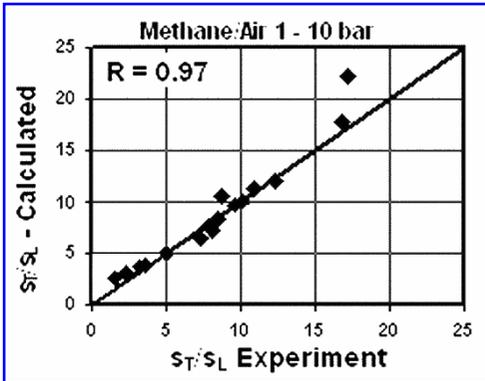
Ethylen-Luft, 1 - 10 bar, 64 Flammen

$$\frac{A_T}{A_L} = 1 + 0.39 \cdot \text{Re}_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

Propan-Luft, 1 - 5 bar, 20 Flammen

$$\frac{A_T}{A_L} = 1 + 0.28 \cdot \text{Re}_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

Modelling of Pressure Influence

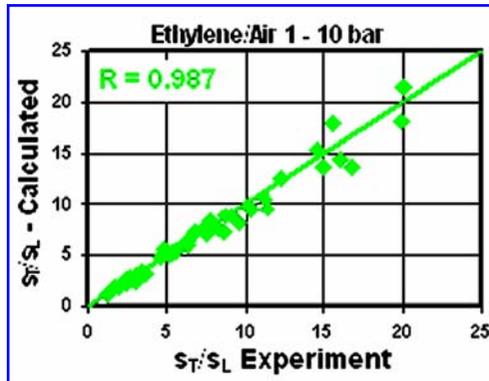


$Le = 1$

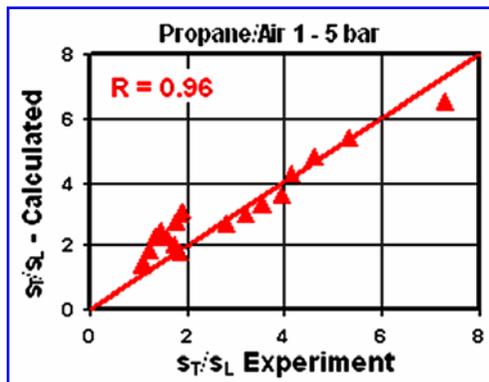
Universell, für magere Methan, Ethylen, Propan-Flammen
(1 - 10 bar)

$$\frac{A_T}{A_L} = 1 + \frac{0.46}{Le} \cdot Re_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

Lewis Zahl ($Le = a / D$)



$Le = 1.2$



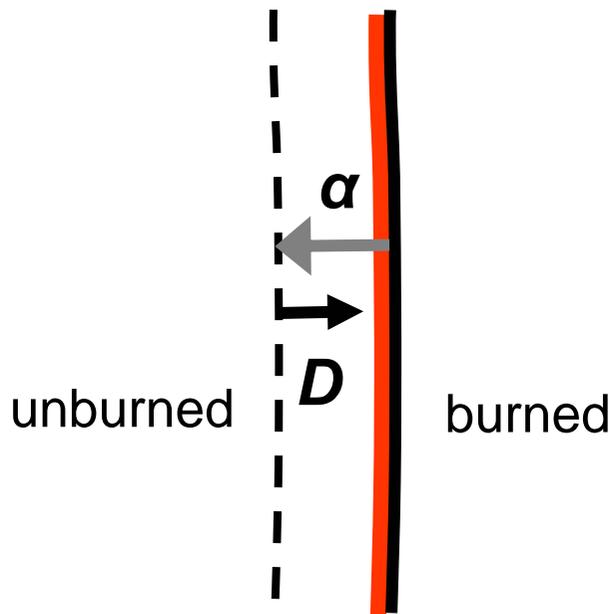
$Le = 1.62$

$$Le = \frac{\alpha}{D} = \frac{\text{thermal diffusivity}}{\text{mass diffusivity (fuel)}}$$

- Why such a big influence of the Lewis number ?
- Molecular diffusion.
- Known for laminar flame instability - but why for highly turbulent combustion ?

$$Le = \frac{\alpha}{D} = \frac{\text{thermal diffusivity}}{\text{mass diffusivity (fuel)}}$$

Laminar plane flame

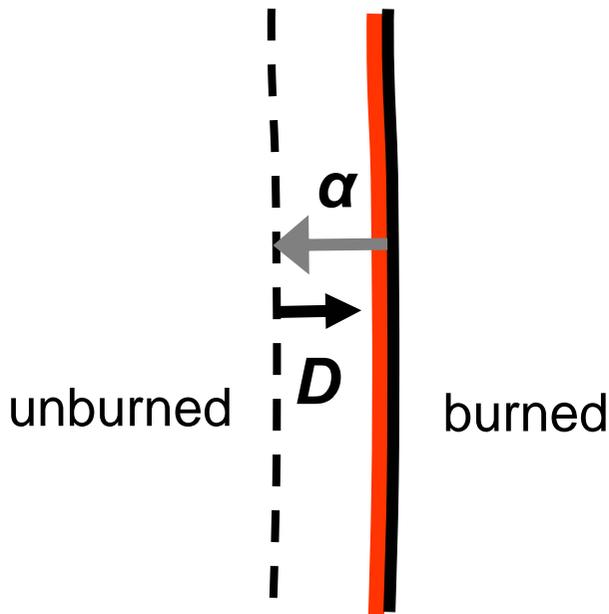


Lewis Number

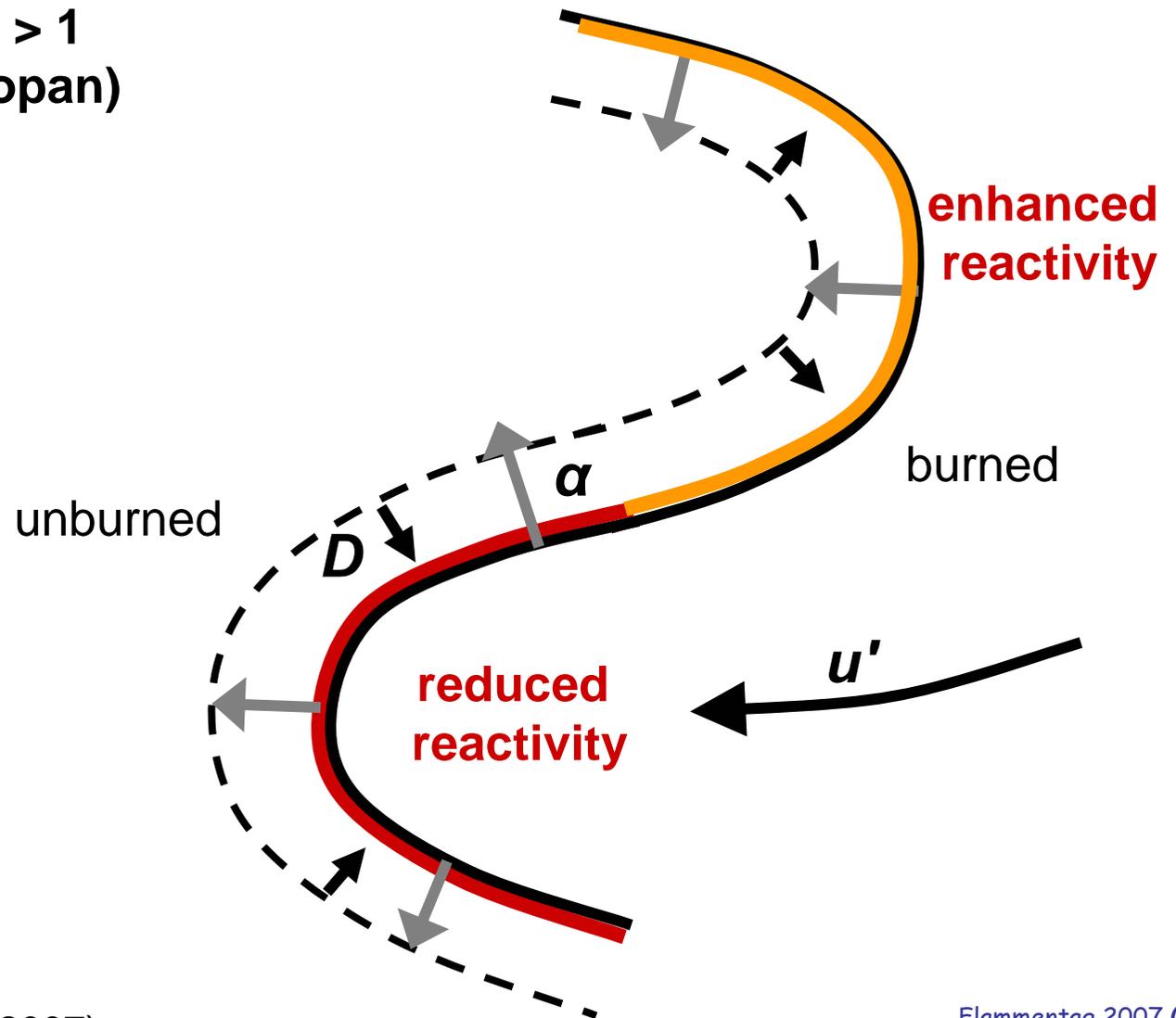
$$Le = \frac{\alpha}{D}$$

Example
 $Le > 1$
(Propan)

Laminar plane flame

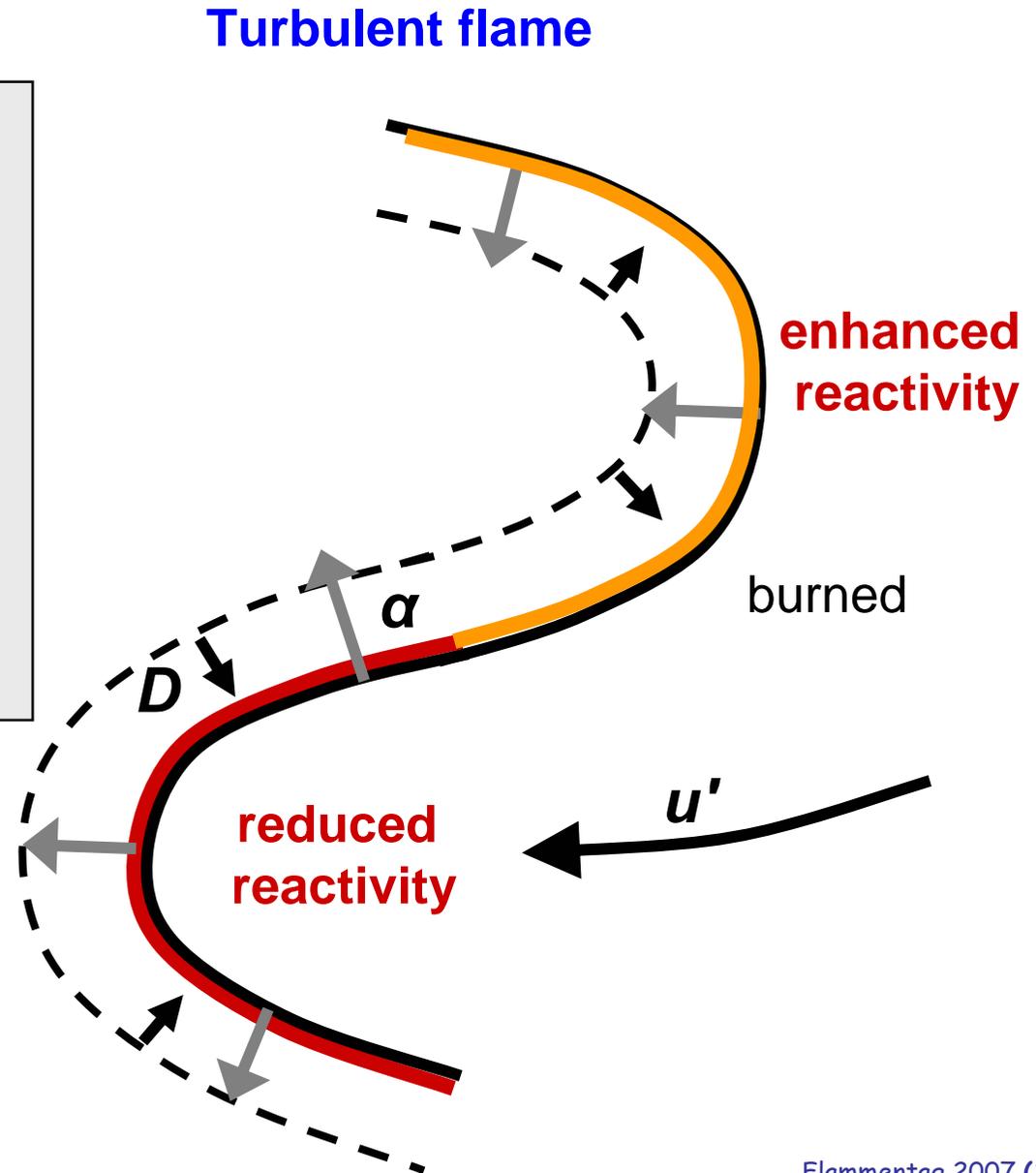


Turbulent flame



Is there Le-influence on "average" reaction rate ???

- Highly turb. flames: pos. and neg. curvature is equally distributed !
- So in first order no Le-influence expected !



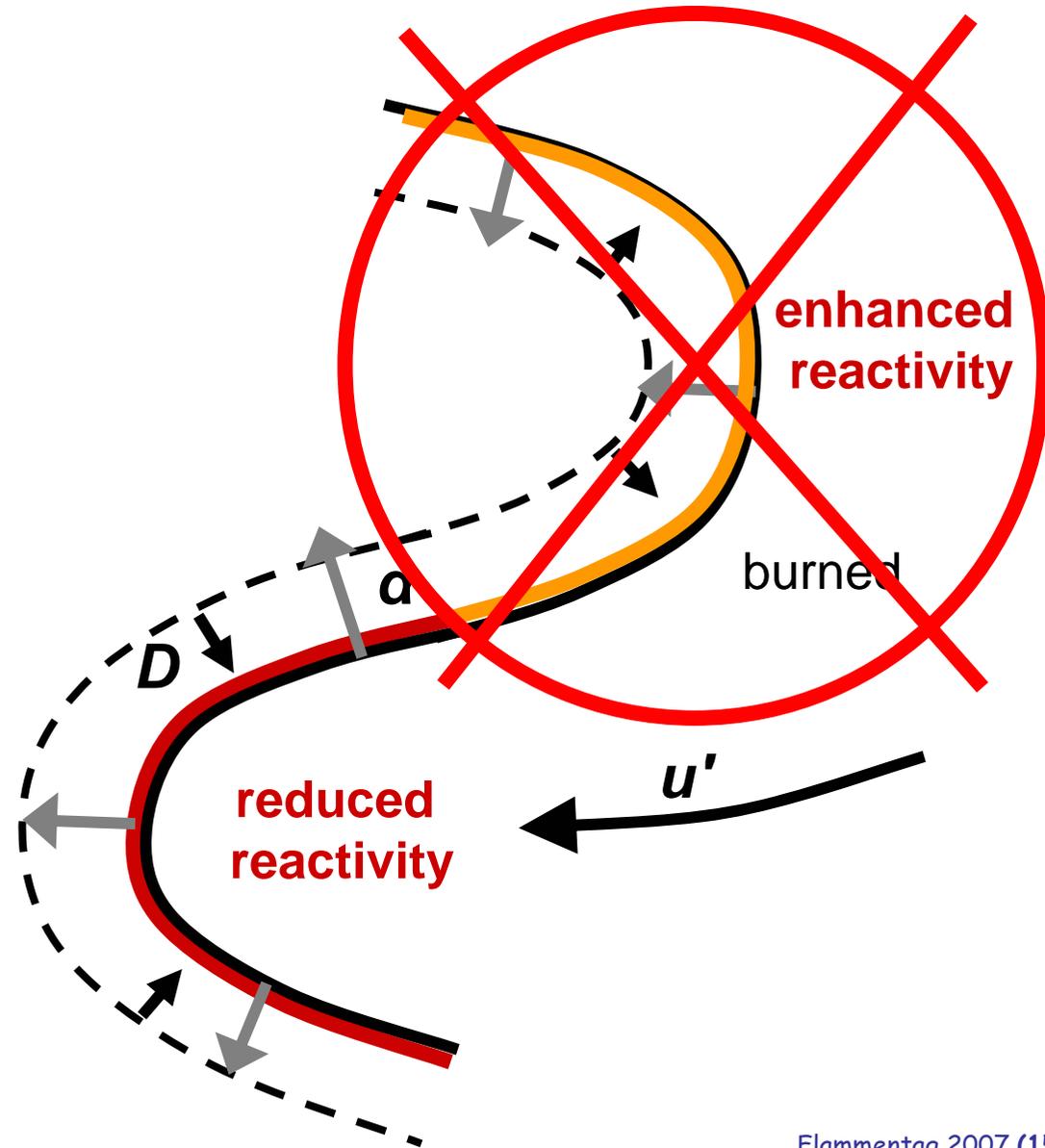
Leading Edge (positive curved part) dominates the flame propagation

(\cong mean reaction rate)

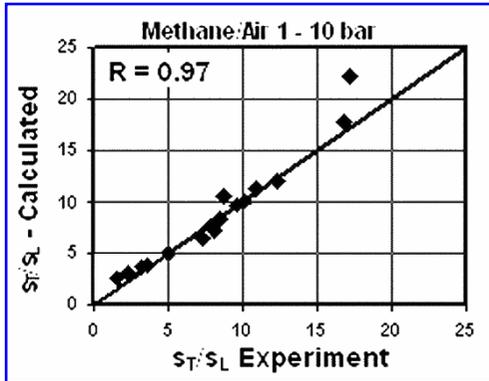
Here fuel diffusion and thermal diffusion eventually unbalanced

Rear part is "only" follow up

$Le > 1$ (Propane):
less reactive as
 $Le = 1$ (Methane)



Modellierung des Druckeinflusses

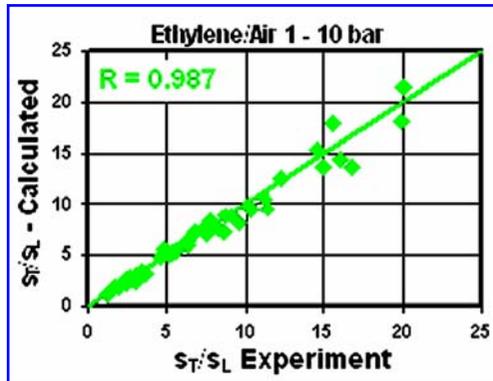


$Le = 1$

Universell, für magere Methan, Ethylen, Propan-Flammen
(1 - 10 bar)

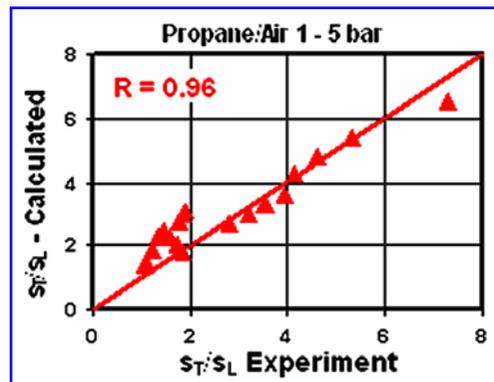
$$\frac{A_T}{A_L} = 1 + \frac{0.46}{Le} Re_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

Lewis Zahl ($Le = a / D$)

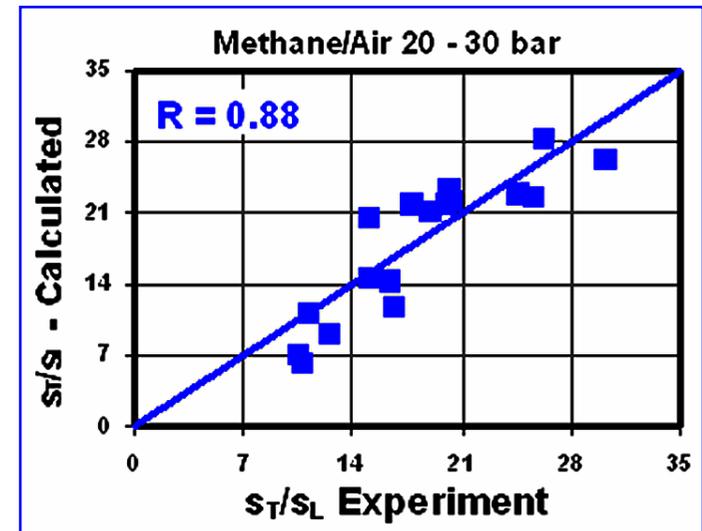


$Le = 1.2$

Extrapolation zu 20 und 30 bar (Vorhersage)



$Le = 1.62$



Berechnung turbulenter Vormischflammen

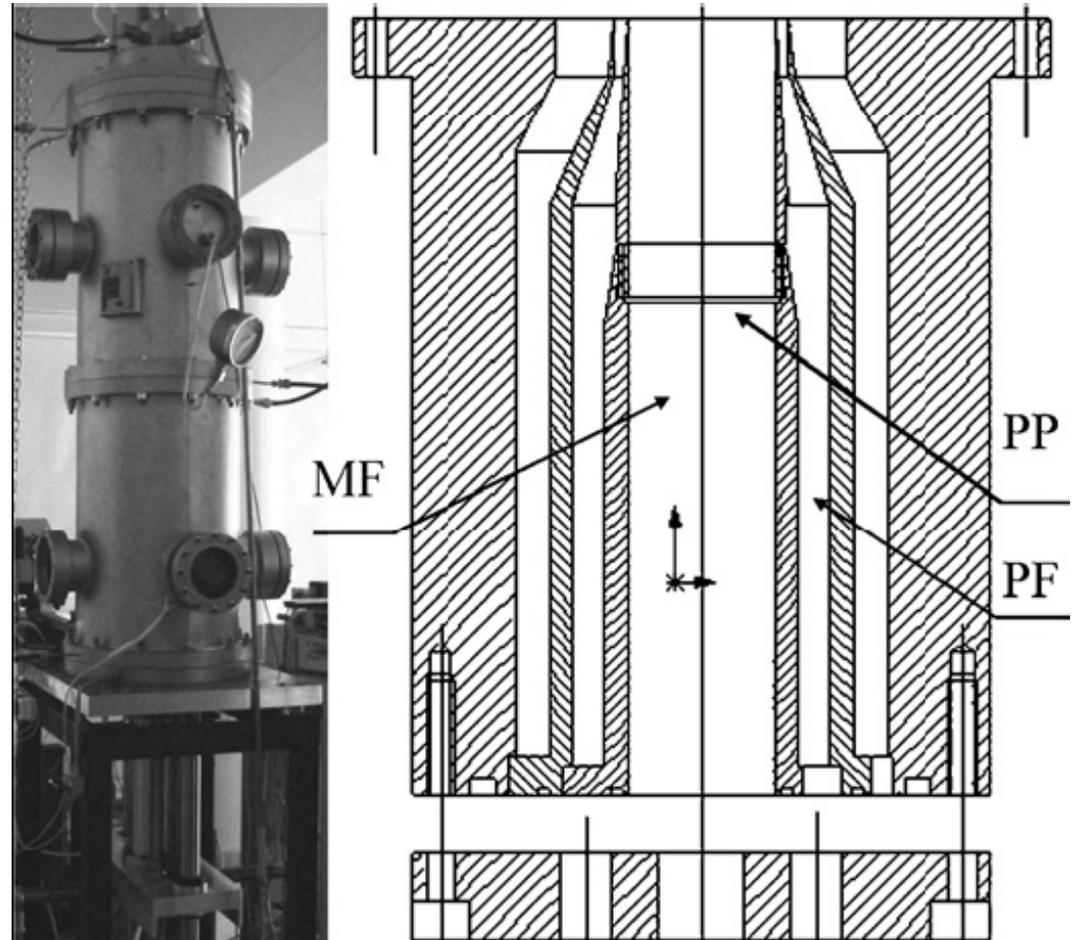
- Reaktionsfortschritts-Modell mit Druck- und Brennstoffeinfluss
- Exkurs Lewis Number
- **Methane/Hydrogen Flames**
- **Effective Lewis Number**
- Zusammenfassung

Hochdruck-Bunsenflammen Orleans

F. Halter, C. Chauveau, I. Gökalp (Univ. Orléans)

Bunsenflamme $D = 25$ mm
H₂-Pilotierung
Turbulenzgitter
Brennkammer 600 mm hoch, $D = 300$
Druck 1, 5, 9 bar

Methane/Hydrogen - Luft
Mager vorgemischt $\Phi = 0,6$
 $U = 2.1$ m/s
 $u' = 0,15$ m/s



$D = 25 \text{ mm}$, Mager vorgemischt $\Phi = 0,6$
 $U = 2.1 \text{ m/s}$, $u' = 0,15 \text{ m/s}$



0.1MPa 0.5MPa 0.9MPa

Pressure Variation

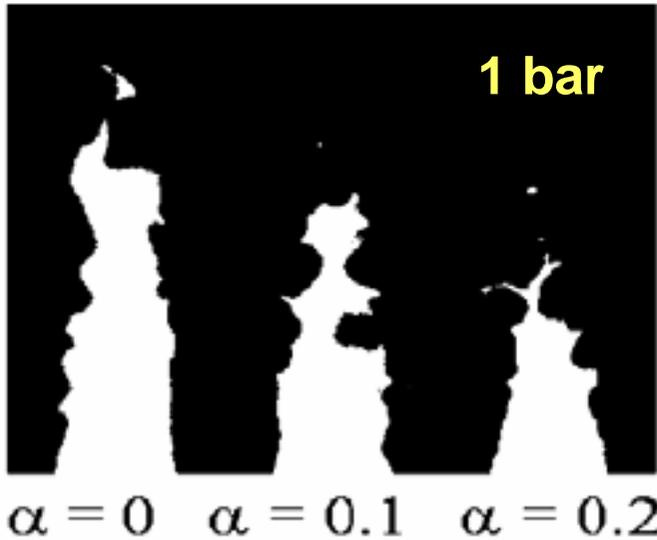


$\alpha = 0$ $\alpha = 0.1$ $\alpha = 0.2$

Hydrogen Addition

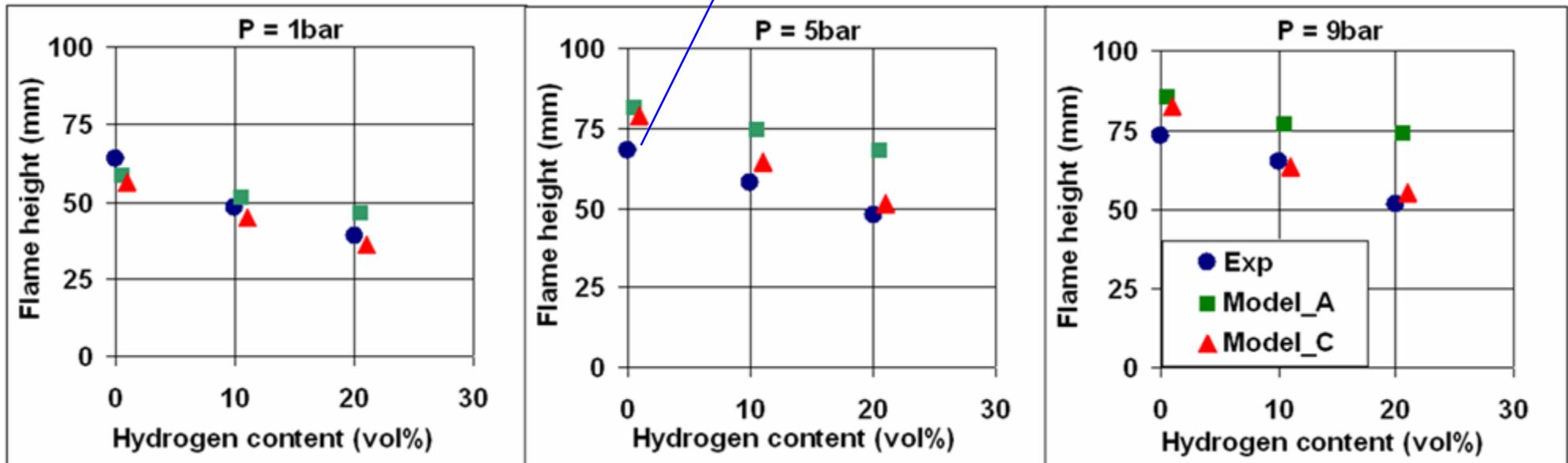
Table 1: Experimental inflow conditions ($U = 2.1$ m/s, $\Phi = 0.6$)

% H ₂	p (MPa)	u' (m/s)	v' (m/s)	l_x (mm)	s_L (m/s)	u'/s_L
0	0.1	0.15	0.12	6.69	0.112	1.36
10	0.1	0.15	0.14	4.30	0.121	1.26
20	0.1	0.16	0.14	3.51	0.132	1.20
0	0.5	0.18	0.14	7.37	0.041	4.36
10	0.5	0.18	0.14	6.97	0.044	4.07
20	0.5	0.19	0.17	6.84	0.048	3.92
0	0.9	0.16	0.11	5.77	0.029	5.32
10	0.9	0.15	0.12	5.45	0.031	4.92
20	0.9	0.16	0.14	5.66	0.033	4.85



Einfluss Wasserstoff-Zufügung

Blau: Experiment (Orléans)



**Modell-Ansatz valid also for Methane/Hydrogen
Flames ???**

$$\frac{A_T}{A_L} = 1 + \frac{0.46}{Le} \cdot \text{Re}_t^{0.25} \cdot \left(\frac{u'}{s_L} \right)^{0.3} \cdot \left(\frac{p}{p_0} \right)^{0.2}$$

**But: Which Lewis Number for
Methane/Hydrogen-Air Flames
???**

**Model A: Effective Lewis Number following Law et al. (2005):
(with q-weighted average)**

$$q (Le^* - 1) = q_1 (Le_1 - 1) + q_2 (Le_2 - 1) \quad (\text{with } q = \text{heat release})$$

Lewis Number

$$Le_{H_2} = 0.29$$

$$Le_{CH_4} = 0.955$$

gives

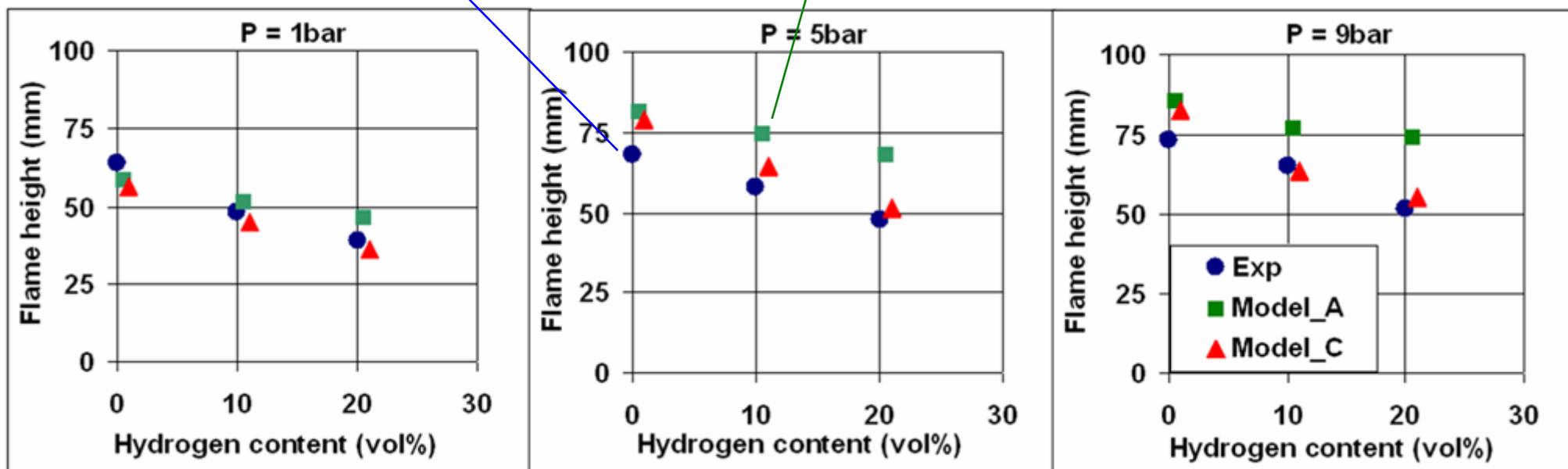
$$Le^* = 0.955 \quad \text{at } 0 \% H_2$$

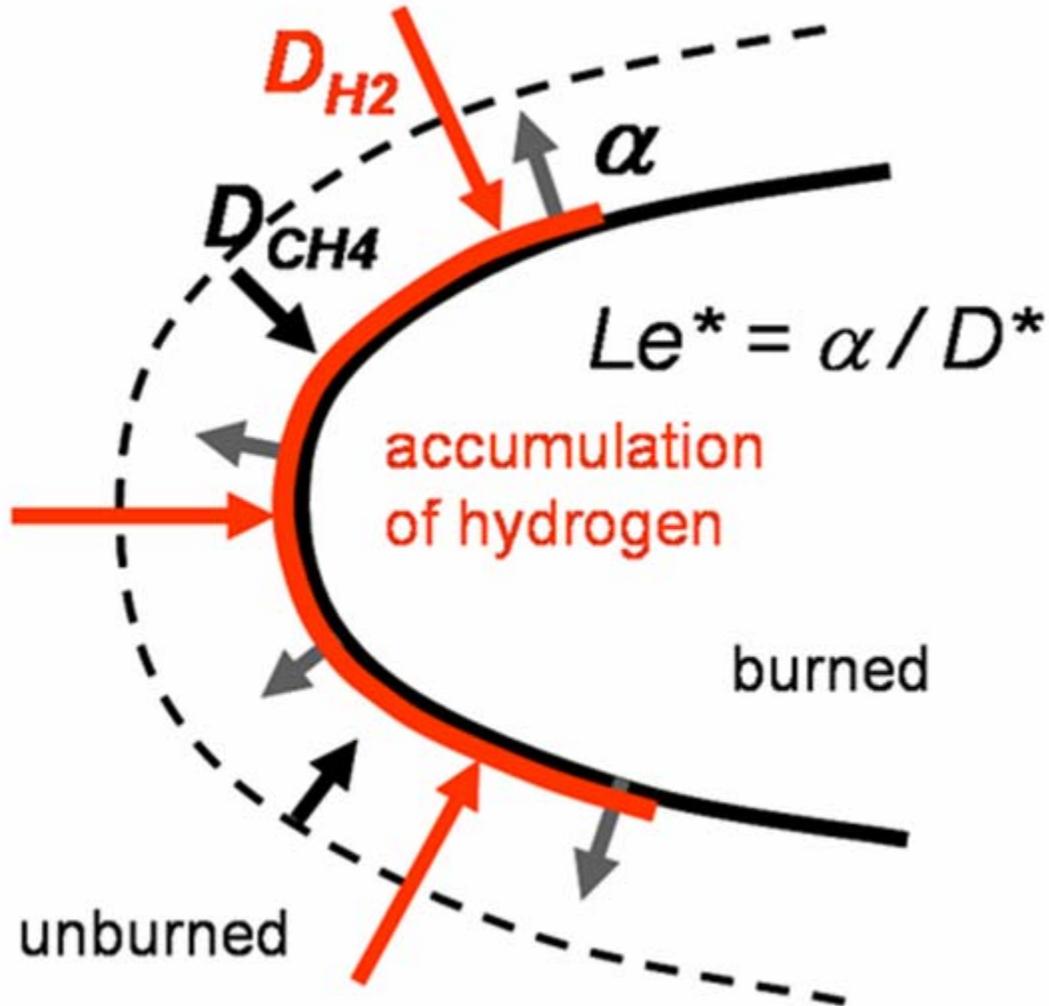
$$Le^* = 0.935 \quad \text{at } 10 \% H_2$$

$$Le^* = 0.918 \quad \text{at } 20 \% H_2$$

Blue:
Experiment

Green: Model A:
Effective Lewis Number foll. Law et al.





Effective Lewis Number Le^*
following Dinkelacker:

$$Le^* = \alpha / D^*$$

$$D^* = x_{H_2} D_{H_2} + x_{CH_4} D_{CH_4}$$

Dinkelacker et al., Dt. Flammentag (2007)

Model C: Effective Lewis Number from effective diffusivity D^* (Dinkelacker 2007):

$$Le^* = \alpha / D^*$$

mit α = thermal diffusivity
 x = mole fraction

$$D^* = x_{H_2} D_{H_2} + x_{CH_4} D_{CH_4}$$

→
$$\frac{1}{Le^*} = \frac{D^*}{a} = \frac{x_{CH_4} D_{CH_4}}{a} + \frac{x_{H_2} D_{H_2}}{a} = \frac{x_{CH_4}}{Le_{CH_4}} + \frac{x_{H_2}}{Le_{H_2}}$$

Lewis Number

$$Le_{H_2} = 0.29$$

$$Le_{CH_4} = 0.95$$

gives

$$Le^* = 0.95 \quad \text{at } 0 \% H_2$$

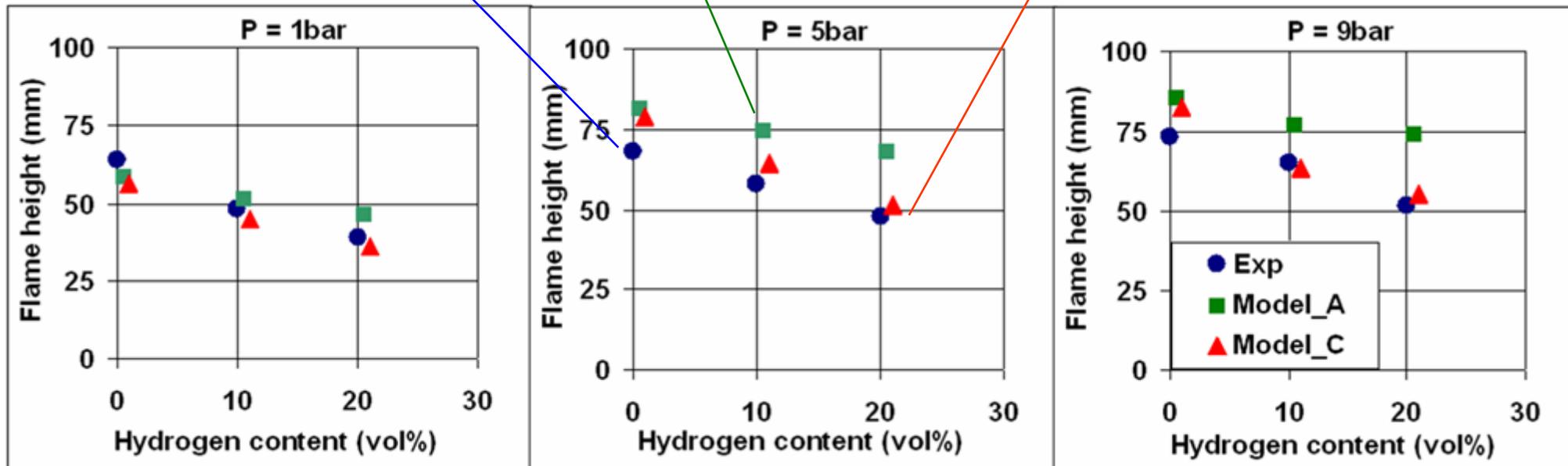
$$Le^* = 0.77 \quad \text{at } 10 \% H_2$$

$$Le^* = 0.65 \quad \text{at } 20 \% H_2$$

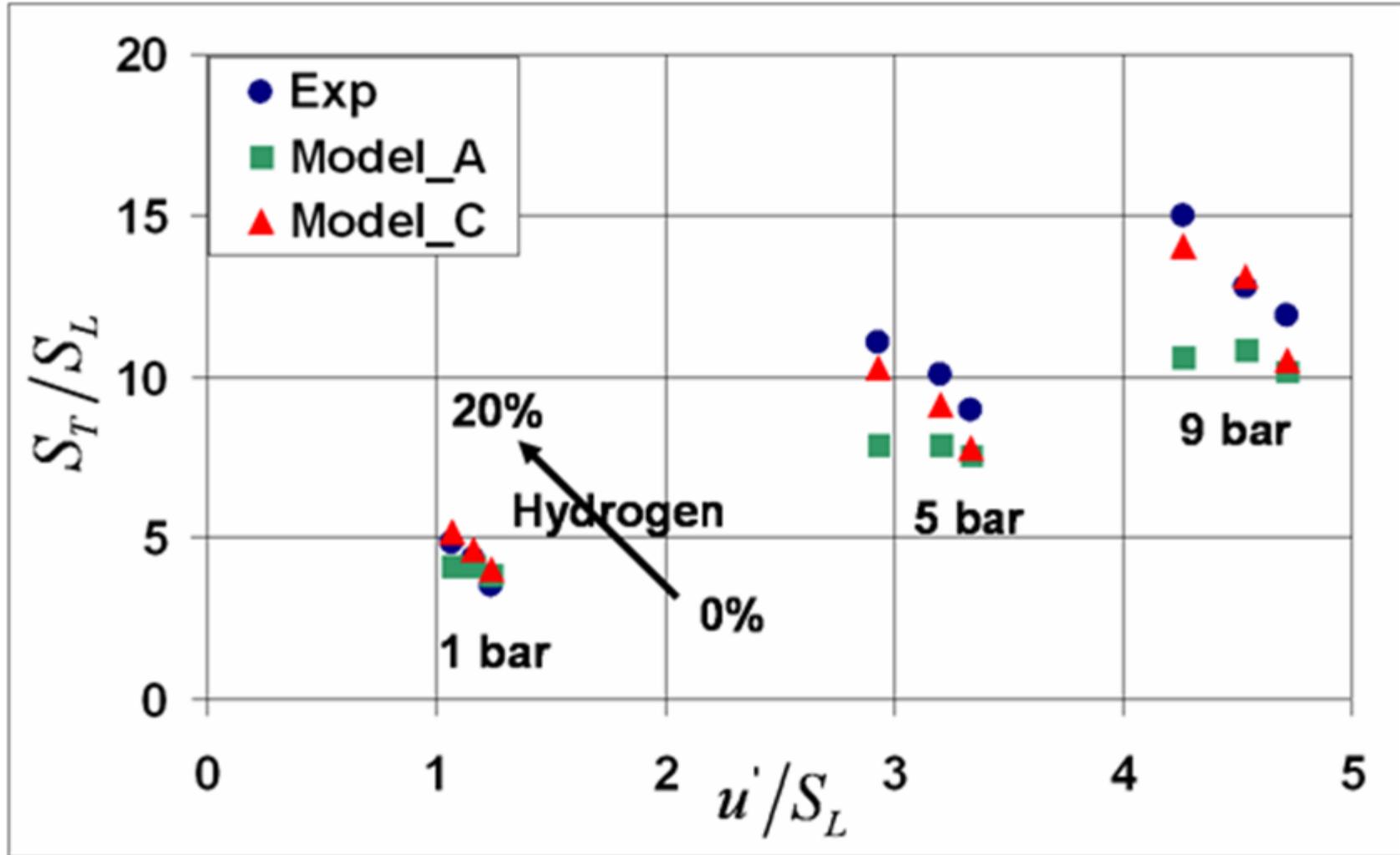
Modell A:
Effective Lewis Number nach Law et al.

Experiment

Modell C:
Effective Lewis Number mittels
Effectiver Diffusivität D^*



Methane/Hydrogen Flame



Numerical calculation of turbulent premixed flames

- **Modelling of mean reaction term**
(TFC Model, good for 1 bar)
- **Calculation of Pressure- + Fuel Influence**
(Muppala-Dinkelacker Model)
- **Le-Influence may be explained with Leading Edge Concept**
- **Calculation of H₂-CH₄ / air flames (1 - 9 bar)**
- **Effective Lewis Number based on Effective Diffusivity of Fuel Mixture**

$$\frac{A_T}{A_L} = 1 + \frac{0.46}{Le} \cdot \text{Re}_t^{0.25} \cdot \left(\frac{u'}{s_L}\right)^{0.3} \cdot \left(\frac{p}{p_0}\right)^{0.2}$$

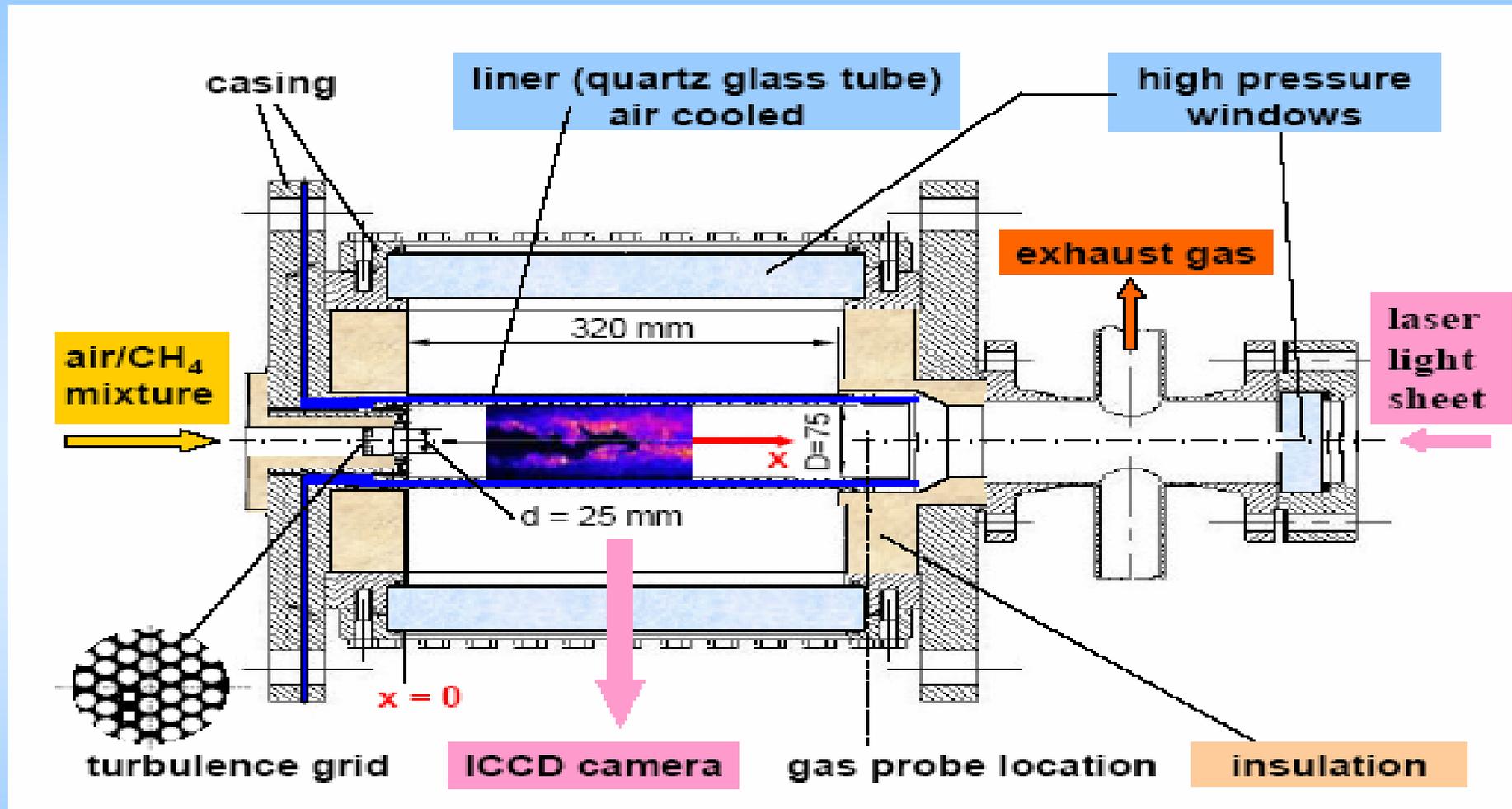
for lean Methane, Ethylene,
Propane-Flames 1 - 10 bar (30 bar)

not for more than
20% hydrogen

$$Le^* = a / D^*$$
$$D^* = x_{H_2} D_{H_2} + x_{CH_4} D_{CH_4}$$

PSI Burner (Griebel, et al.)

5 bar, $\Phi = 0.5$, Methane/Hydrogen - Air



High-pressure combustor (PSI)

PSI Burner (Griebel, et al.)

Methane/Hydrogen - Air, 5 bar, $\Phi = 0.5$

